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Highly efficient non-rare-earth red emitting phosphor for warm white light-emitting diodes

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Mn⁴⁺-activated fluoride compounds, as an alternative to commercial (oxy)nitride phosphors, are emerging as a new class of non-rare-earth red phosphors for high-efficacy warm white LEDs. Currently, it remains a challenge to synthesize these phosphors with high photoluminescence quantum yields through a convenient chemical route. Herein we propose a general but convenient strategy based on efficient cation exchange reaction, which had been originally regarded only effective in synthesizing nano-sized materials before, for the synthesis of Mn⁴⁺-activated fluoride microcrystals such as K₂TiF₆, K₂SiF₆, NaGdF₄ and NaYF₄. Particularly we achieve a photoluminescence quantum yield as high as 98% for K₂TiF₆:Mn⁴⁺. By employing it as red phosphor, we fabricate a high-performance white LED with low correlated colour temperature (3,556 K), high-colour-rendering index ($R_a = 81$) and luminous efficacy of 116 lm W⁻¹. These findings show great promise of K₂TiF₆:Mn⁴⁺ as a commercial red phosphor in warm white LEDs, and open up new avenues for the exploration of novel non-rare-earth red emitting phosphors.

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